

## Isotope Labeling Studies Reveal the Order of Oxygen Incorporation into the Tryptophan Tryptophylquinone Cofactor of Methylamine Dehydrogenase

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An increasing number of enzymes of eukaryotic and prokaryotic origin have been shown to contain novel cofactors derived from endogenous amino acids.<sup>1</sup> Many of these protein-derived cofactors are formed autocatalytically.<sup>1</sup> The tryptophan tryptophylquinone (TTQ) cofactor of methylamine dehydrogenase (MADH) is different in that formation of active enzyme requires co-expression of four processing genes,<sup>2,3</sup> including *mauG*, which encodes a di-heme *c*-type cytochrome with unusual properties.<sup>4,5</sup>

TTQ is derived from two  $\beta$  subunit residues of MADH,  $\beta$ W57 and  $\beta$ W108 in *Paracoccus denitrificans*.<sup>6</sup> Two oxygens are added to  $\beta$ W57 at positions C6 and C7, and a cross-link is formed between C4 of  $\beta$ W57 and C2 of  $\beta$ W108 (Figure 1). X-ray crystallographic studies have demonstrated that the C6 position of  $\beta$ W57 is the site of nucleophilic attack by substrate during MADH turnover, displacing the oxygen at C6 to form a Schiff's base.<sup>7</sup>

Previous studies have revealed that the biogenesis of TTQ is a multistep process that occurs after the MADH  $\beta$  subunit is translocated into the periplasm.<sup>2,3,8–10</sup> The earliest intermediate in TTQ synthesis so far observed is a species lacking any modifications to either  $\beta$ W57 or  $\beta$ W108, but with a full complement of six intrasubunit disulfide bonds.<sup>10</sup> The second intermediate that has been isolated is monohydroxylated at  $\beta$ W57 and lacks a cross-link to  $\beta$ W108.<sup>9</sup> The mechanism of this first hydroxylation is poorly understood; however, it is known to require the MADH active site residue  $\beta$ D76.<sup>10</sup> As  $\beta$ D76 is located close to the C6 position of TTQ in the mature enzyme, it seemed likely that the first oxygen was incorporated into  $\beta$ W57 at this position.<sup>10</sup> The second hydroxylation, cross-link formation, and oxidation to quinone are catalyzed in an O<sub>2</sub> and reducing-equivalent-dependent manner by MauG.<sup>4,9,11</sup>

To further probe oxygen incorporation into TTQ, the monohydroxylated MADH biosynthetic intermediate lacking the cross-link was incubated in vitro with purified MauG, NADH, and either isotopically labeled water (H<sub>2</sub><sup>18</sup>O, 95%, Cambridge Isotope Laboratories) or O<sub>2</sub> (<sup>18</sup>O<sub>2</sub>, 99%, Isotec Inc.) for 24 h. Mass spectrometry was used to assess the formation of TTQ as well as to determine whether <sup>18</sup>O had been incorporated.

Initial experiments, in which mature MADH was incubated in buffer containing H<sub>2</sub><sup>18</sup>O, confirmed previous studies<sup>12</sup> which reported that MADH contains only one exchangeable carbonyl (Figure 2, green trace), consistent with the crystal structure in which the C6 carbonyl is exposed to solvent while the C7 carbonyl is hydrogen-bonded to a backbone amide.<sup>6</sup> Further incubation of singly

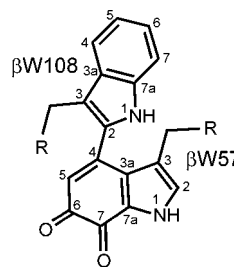


Figure 1. Structure of mature TTQ.

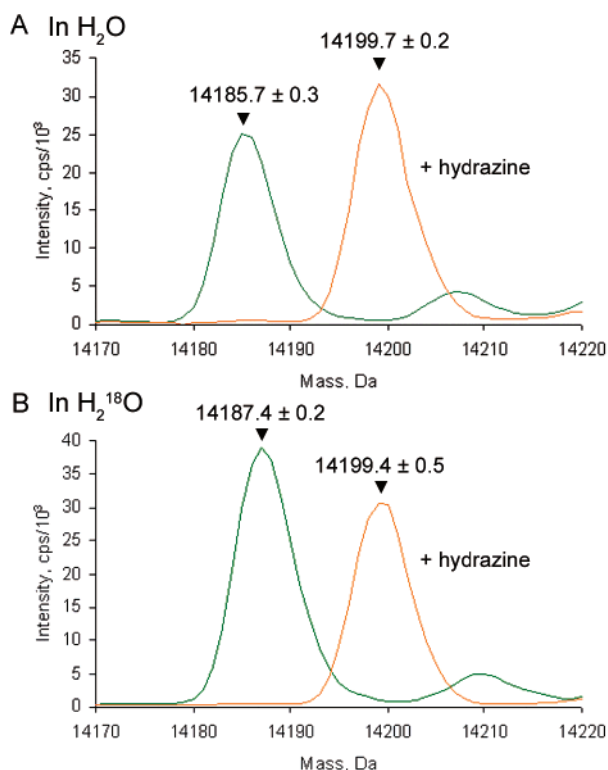


Figure 2. Mature MADH  $\beta$  subunit contains one exchangeable carbonyl at the C6 position of  $\beta$ W57. Sample preparation and mass spectrometry analysis were as previously described.<sup>9</sup> Shown are deconvoluted mass spectra of the mature MADH  $\beta$  subunit. Green lines are the spectra after overnight incubation in buffer containing (A) H<sub>2</sub><sup>16</sup>O and (B) H<sub>2</sub><sup>18</sup>O; orange lines are the spectra after the further addition of a 5-fold excess of hydrazine to each sample.

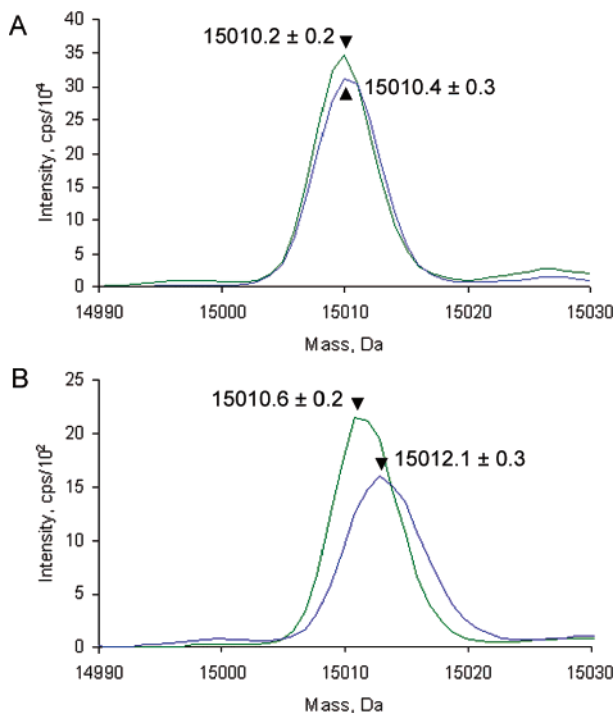
<sup>18</sup>O-labeled and unlabeled MADH with hydrazine, which displaces the oxygen at C6 of  $\beta$ W57 forming a covalent bond,<sup>7</sup> results in loss of the <sup>18</sup>O label (Figure 2, orange trace). This means that any oxygen atom added to the C6 position of TTQ during synthesis will exchange with solvent upon cofactor maturation.

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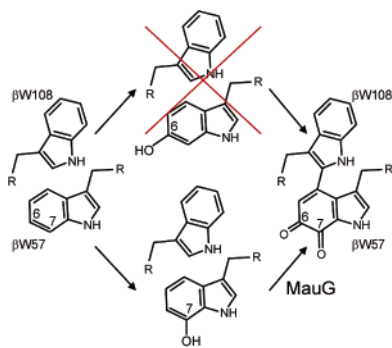
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**Figure 3.** MauG catalyzes insertion of the second oxygen at the C6 position of  $\beta$ W57. Reaction mixtures contained 10  $\mu$ M MADH  $\beta$  subunit monohydroxylated biosynthetic intermediate,<sup>9</sup> 1  $\mu$ M MauG, and 60  $\mu$ M NADH. (A) The reaction was carried out in the presence of  $^{16}\text{O}_2$  (green) or  $^{18}\text{O}_2$  (blue). (B) The reaction was carried out in buffer containing either  $\text{H}_2^{16}\text{O}$  (green) or  $\text{H}_2^{18}\text{O}$  (blue). Sample preparation and mass spectrometry analysis were as previously described.<sup>9</sup> The larger mass of the mature subunit relative to that in Figure 2 is because the biosynthetic intermediate used as substrate in this experiment included a 6 $\times$  histidine C-terminal tag.



**Figure 4.** Refined mechanism of TTQ maturation.

The MauG-dependent TTQ biosynthetic reaction requires  $\text{O}_2$ .<sup>9,11</sup> When the reaction was carried out in the presence of unlabeled water and  $^{18}\text{O}_2$ , mass spectrometry showed that no label was present in the mature TTQ cofactor (Figure 3A). This observation indicates that the MauG-dependent addition of oxygen does not occur at the nonexchangeable C7 position (Figure 4). If it did, then  $^{18}\text{O}$  would be present in the mature TTQ. This result strongly suggests that the second oxygen is added at C6 and then rapidly exchanged with  $\text{H}_2\text{O}$ . The only other possibility is that the second oxygen could be derived from solvent following  $\text{O}_2$ -dependent substrate activation. To test this, the MauG-dependent biosynthetic reaction was carried out in the presence of  $\text{H}_2^{18}\text{O}$ . Mass spectrometry showed that a

single  $^{18}\text{O}$  had been incorporated into the resulting TTQ (Figure 3B). If the MauG-dependent second oxygen is derived from solvent and added at the nonexchangeable C7 position, then two  $^{18}\text{O}$  atoms should be present, since the oxygen at position C6 of  $\beta$ W57 would rapidly exchange with solvent upon cofactor maturation. Thus, these results confirm that the first oxygen is added to C7 during TTQ biogenesis and that the second MauG-dependent oxygenation occurs at C6.

These results suggest that the structure of the initial biosynthetic intermediate of MADH with no oxygens added to  $\beta$ W57 differs from that of the mature enzyme, despite the fact that the six disulfide cross-links of the  $\beta$  subunit are already formed. In particular, it seems likely that the position of  $\beta$ D76 relative to  $\beta$ W57 is altered, as this residue is required for the first hydroxylation,<sup>10</sup> shown here to be at the C7 position of  $\beta$ W57 rather than C6, which is much closer to  $\beta$ D76 in mature MADH. A different pre-TTQ maturation  $\beta$  subunit conformation is also consistent with our previous finding that the  $\alpha\beta$  subunit interactions of MADH are relatively weak until TTQ biogenesis is complete.<sup>9,13</sup>

These data have enabled us to refine our model of TTQ biogenesis (Figure 4). They indicate that, after translocation into the periplasm and formation of the six disulfide linkages, the first MauG-independent hydroxylation of  $\beta$ W57 occurs specifically at the C7 position in a  $\beta$ D76-dependent manner.<sup>6</sup> MauG then catalyzes insertion of the second oxygen at the C6 position of  $\beta$ W57, formation of a cross-link to  $\beta$ W108, and oxidation to a quinone.<sup>8</sup> Upon completion of TTQ maturation, the quinone carbonyl at position C6 of  $\beta$ W57 exchanges rapidly with solvent. Therefore, it is not possible to determine from these data the source of the oxygen incorporated by MauG, although other recent work favors a P450-like mechanism in which MauG inserts an oxygen derived from  $\text{O}_2$ .<sup>11</sup>

**Acknowledgment.** This work was supported by NIH Grants GM-41574 (V.L.D.) and GM-66569 (C.M.W.). Mass spectrometry studies were carried out at the Center for Mass Spectrometry and Proteomics at The University of Minnesota.

## References

- Okeley, N. M.; van der Donk, W. A. *Chem. Biol.* **2000**, *7*, R159–171.
- van der Palen, C. J.; Slotboom, D. J.; Jongejan, J. A.; Reijnders, W. N.; Harms, N.; Duine, J. A.; van Spanning, R. J. *Eur. J. Biochem.* **1995**, *230*, 860–871.
- van der Palen, C. J.; Reijnders, W. N.; de Vries, S.; Duine, J. A.; van Spanning, R. J. *Antonie Van Leeuwenhoek* **1997**, *72*, 219–228.
- Wang, Y.; Li, X.; Jones, L. H.; Pearson, A. R.; Wilmot, C. M.; Davidson, V. L. *J. Am. Chem. Soc.* **2005**, *127*, 8258–8259.
- Wang, Y.; Graichen, M. E.; Liu, A.; Pearson, A. R.; Wilmot, C. M.; Davidson, V. L. *Biochemistry* **2003**, *42*, 7318–7325.
- Chen, L.; Doi, M.; Durlay, R. C.; Chistoserdov, A. Y.; Lidstrom, M. E.; Davidson, V. L.; Mathews, F. S. *J. Mol. Biol.* **1998**, *276*, 131–149.
- Huizinga, E. G.; van Zanten, B. A.; Duine, J. A.; Jongejan, J. A.; Huitema, F.; Wilson, K. S.; Hol, W. G. *Biochemistry* **1992**, *31*, 9789–9795.
- Chistoserdov, A. Y.; Chistoserdova, L. V.; McIntire, W. S.; Lidstrom, M. E. *J. Bacteriol.* **1994**, *176*, 4052–4065.
- Pearson, A. R.; De la Mora-Rey, T.; Graichen, M. E.; Wang, Y.; Jones, L. H.; Marimanikkupam, S.; Agger, S. A.; Grimsrud, P. A.; Davidson, V. L.; Wilmot, C. M. *Biochemistry* **2004**, *43*, 5494–5502.
- Jones, L. H.; Pearson, A. R.; Tang, Y.; Wilmot, C. M.; Davidson, V. L. *J. Biol. Chem.* **2005**, *280*, 17392–17396.
- Li, X.; Jones, L. H.; Pearson, A. R.; Wilmot, C. M.; Davidson, V. L. Unpublished.
- Backes, G.; Davidson, V. L.; Huitema, F.; Duine, J. A.; Sanders-Loehr, J. *Biochemistry* **1991**, *30*, 9201–9210.
- Pearson, A. R.; Jones, L. H.; Higgins, L.; Ashcroft, A. E.; Wilmot, C. M.; Davidson, V. L. *Biochemistry* **2003**, *42*, 3224–3230.

JA064466E